

## Sulfur Hexafluoride (SF<sub>6</sub>) Awareness

No. 2009-01

August 2009

### PURPOSE

This bulletin provides information on the environmental and sustainable management of sulfur hexafluoride (SF<sub>6</sub>) at DOE sites. SF<sub>6</sub> is an extremely potent greenhouse gas.

### BACKGROUND

Sulfur hexafluoride is a colorless, odorless, non-toxic synthetic gas commonly used to insulate electrical equipment. Equipment such as circuit breakers, current-interruption equipment, transmission lines, transformers, and substations account for over 80% of the annual use of SF<sub>6</sub> in the United States. Atmospheric emissions occur during equipment maintenance or leaks. In addition to electrical equipment, SF<sub>6</sub> may be intentionally released directly to the air in small amounts as a tracer compound for air dispersion studies.

### WHAT ARE THE HAZARDS?

An extremely potent greenhouse gas (GHG), the release of one metric ton (tonne) of SF<sub>6</sub> is equivalent to 23,900 tonnes of carbon dioxide (CO<sub>2</sub>). Small releases, therefore, significantly contribute to site GHG emissions.

The material safety data sheet (MSDS) for SF<sub>6</sub> indicates that it is non-combustible and non-flammable. However, standard safety procedures for handling pressurized gas cylinders and equipment should be followed (see HSS Safety Bulletin 2007-01, [http://www.hss.energy.gov/CSA/csp/safety\\_bulletins/SB\\_2007-01.pdf](http://www.hss.energy.gov/CSA/csp/safety_bulletins/SB_2007-01.pdf)). In the event of fire, personnel should be evacuated and fire-fighters should wear self-contained breathing apparatus as SF<sub>6</sub> may create hazardous gases including disulfur decafluoride (S<sub>2</sub>F<sub>10</sub>).

### CONTROLLING HAZARDS THROUGH LIFECYCLE MANAGEMENT

Currently, there is no direct gaseous substitute for SF<sub>6</sub>, so emission reductions require controlling releases. A lifecycle approach to SF<sub>6</sub> management, including a focus on tracking inventory, leak detection and repair, and proper disposition including recycling, is a recommended best practice.

A **tracking program** should include a complete inventory of SF<sub>6</sub> including purchases, on-site storage, equipment capacity (nameplate), and disbursements. Tracking practices include the following:

- Inventory of SF<sub>6</sub> cylinder locations from arrival to departure. Include the location of all equipment and storage areas where SF<sub>6</sub> is found. One cylinder (115 lbs. of SF<sub>6</sub>) is the equivalent of 1,250 tonnes of CO<sub>2</sub>.

- Use of SF<sub>6</sub> on-site, as equipment requiring frequent "topping-off" may need maintenance or replacement.
- Quantity of SF<sub>6</sub> consumed, as partially filled containers shipped offsite would otherwise be assumed full.

**Leak detection** is essential to minimizing releases.

According to EPA, 10% of circuit breakers leak and 15% of these are readily repairable. The majority of leaks occur at gas mechanisms, bushings, and gas tanks.

Assess usage and inventory data to identify equipment that may be leaking. Maintenance and replacement can be prioritized according to the severity of leaks.

Leak identification can be aided by low-gas density alarms, handheld halogen detectors, and laser cameras. Detection techniques include spraying non-electrical or de-energized equipment with a solution of soap and water and use of microprocessor density monitors on equipment.

**Proper disposition including recycling** will reduce atmospheric emissions. During maintenance, evacuate SF<sub>6</sub> into portable cylinders instead of venting to the atmosphere. Recaptured SF<sub>6</sub> can be re-used on-site or shipped to a recycler. Both air and water can degrade the purity of SF<sub>6</sub> and are not easily removed. An on-site SF<sub>6</sub> recapture and reuse program requires the following:

- Coverage under the site health and safety program,
- Resources and procedures to recover, store, and test the purity of SF<sub>6</sub> for on-site reuse,
- A minimum acceptable purity standard for SF<sub>6</sub> on-site reuse, and
- Disposition to an authorized SF<sub>6</sub> dealer/recycler if SF<sub>6</sub> is not reusable on-site.

### ADDITIONAL SOURCES OF INFORMATION

Contact your site Environmental, Safety, and Health staff or Jeff Eagan at 202-586-4598 or [jeff.eagan@hq.doe.gov](mailto:jeff.eagan@hq.doe.gov).

### SUMMARY

SF<sub>6</sub> is a potent GHG. SF<sub>6</sub> emissions can be reduced through tracking, leak detection, maintenance, and recycling.



Glenn S. Podonsky  
Chief Health, Safety and Security Officer  
Office of Health, Safety and Security



## PREVENT EVENTS

Learning from Industry Experience

**PREVENT EVENTS is intended for use by personnel during morning meetings, pre-job briefings, and work unit meetings to communicate key industry experience.**

### Management:

1. Is there an inventory of SF<sub>6</sub> containing equipment?
2. Have we developed statistical survey methods and plans to assess suspect equipment for SF<sub>6</sub> releases?
3. Are our procurement and chemical inventory tracking systems setup to help minimize the size and quantity of pressurized SF<sub>6</sub> cylinders at our site?
4. Are our environment, safety and health people keeping our utilities and facilities management people informed of the latest developments in environmental sustainable management of SF<sub>6</sub>?
5. What training have we provided our workers on the characterization of SF<sub>6</sub>-containing equipment and reporting of SF<sub>6</sub> releases?
6. What environment, safety and health programs are required to institute a recovery and reuse program for SF<sub>6</sub>?
7. Have we calculated the impact of SF<sub>6</sub> on our site GHG inventory?
8. Are we tracking SF<sub>6</sub> disposition practices?

### Supervisors and Workers:

1. Have all employees for the job been properly trained to identify SF<sub>6</sub> leaks from electrical equipment?
2. What should we do with leaking cylinders or equipment?
3. Have employees for the job been trained on recovery and reuse procedures for SF<sub>6</sub>?
4. What is the minimum acceptable purity standard for SF<sub>6</sub> in the equipment?
5. Does the facility have resources to recover, store, and test the purity of SF<sub>6</sub>, or will a specialty contractor be required?
6. Are there special equipment or control measures to minimize accidental release of SF<sub>6</sub> during evacuation?
7. How much vacuum must be generated for SF<sub>6</sub> recovery? Does the manufacturer specify evacuation procedures?
8. Is an oil compressor required for reclamation, or can an oil-less compressor be used to avoid contamination?
9. Are SF<sub>6</sub> disposition practices clearly defined and appropriate to minimize releases?

